

Periodic Table of the Elements



11.01

H

Hydrogen

3.04

Li

Lithium

6.94

Be

Beryllium

22.99

Na

Sodium

24.31

Mg

Magnesium

39.10

K

Potassium

40.08

Ca

Calcium

87.62

Rb

Rubidium

132.91

Cs

Cesium

223.02

Fr

Francium

12

Si

Silicon

14

14.01

Symbol

28.09

Atomic weight

14

Atomic number

silicon

name

black

solid

blue

liquid

red

gas

white

synthetically prepared

near ambient temp

alkali metals

alkaline earth metals

transitional metals

other metals

nonmetals

noble gases

2

He

Helium

5.01

B

Boron

12.01

C

Carbon

14.01

N

Nitrogen

16.00

O

Oxygen

18.99

F

Fluorine

20.18

Ne

Neon

26.98

Al

Aluminum

28.09

Si

Silicon

30.97

P

Phosphorus

32.06

S

Sulfur

35.45

Cl

Chlorine

39.96

Ar

Argon

69.72

Ga

Gallium

72.64

Ge

Germanium

74.92

As

Arsenic

78.96

Se

Selenium

79.90

Br

Bromine

83.80

Kr

Krypton

114.82

In

Indium

118.71

Sn

Tin

121.76

Sb

Antimony

127.60

Te

Tellurium

126.91

I

Iodine

131.29

Xe

Xenon

186.21

Tl

Thallium

207.2

Pb

Lead

208.98

Bi

Bismuth

209

Po

Polonium

210

At

Astatine

210

Rn

Radon

(113)

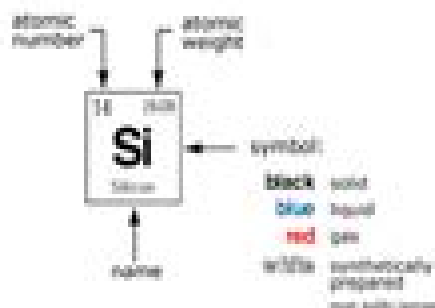
(114)

(115)

(116)

(117)

(118)



| | | | | | | | | | | | | | |
|---------------------|--------------------------|-----------------------|------------------------|-----------------------|-----------------------|------------------------|-----------------------|-------------------------|-------------------------|----------------------|--------------------------|-----------------------|-------------------------|
| 58 Ce Cerium | 59 Pr Praseodymium | 60 Nd Neodymium | 61 Pm Promethium | 62 Sm Samarium | 63 Eu Europium | 64 Gd Gadolinium | 65 Tb Terbium | 66 Dy Dysprosium | 67 Ho Holmium | 68 Er Erbium | 69 Tm Thulium | 70 Yb Ytterbium | 71 Lu Lutetium |
| 90 Th Thorium | 91 Pa Protactinium | 92 U Uranium | 93 Np Neptunium | 94 Pu Plutonium | 95 Am Americium | 96 Cm Curium | 97 Bk Berkelium | 98 Cf Californium | 99 Es Einsteinium | 100 Fm Fermium | 101 Md Mendelevium | 102 No Nobelium | 103 Lr Lawrencium |

Physical Science Lab Periodic Table

Ben Highmore



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Exploring Physical Science in the Laboratory John T. Salinas, 2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging Exploring Physical Science in the Laboratory guides students through the mysteries of the observable world and helps them develop a clear understanding of challenging concepts **Science Mysteries Explained** Anthony Fordham, 2014-06-05 Ideal for the armchair science enthusiast Idiot's Guides Science Mysteries Explained takes a question answer based approach to teach readers a wide variety of topics in Earth Science Life Science Chemistry Physics and Cosmology Using helpful full color illustrations and expert information this book features 130 fascinating questions and answers to satisfy any scientist wannabe

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explained clearly and concisely Understanding the Periodic Table of Chemical Elements is critical for success in the chemistry classroom and laboratory In today s classroom students not only need to understand the properties of the chemical elements but how these elements play such an integral role in industry the earth and the environment and in modern life No resource provides a better introduction than Robert Krebs s The History and Use of Our Earth s Chemical Elements In this thoroughly revised edition with extensive new and updated examples on the use of the chemical elements the elements are examined within their groups enabling students to make connections between elements of similar structure In addition the discovery and history of each element from those known from ancient times to those created in the modern laboratory is explained clearly and concisely In addition to the handy Guide to the Chemical Elements that comprises the bulk of the work The History and Use of Our Earth s Chemical Elements includes other useful features Introductory material on the basics of chemistry and the Periodic Table Appendices on the discoverers of the chemical elements A glossary of words commonly used in chemistry and chemical engineering A complete bibliography of useful resources including websites All of this information makes The History and Use of Our Earth s Chemical Elements the ideal one volume resource for understanding the importance of the chemical elements

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