

Use your notebook to complete all of the following review questions.

Please note that your completed notebook, corrected quizzes and unit review sheets are the **BEST** study tools. These questions will only help to reinforce some of the exam topics.

1. Draw a line continuum to show the metric prefixes in order from largest (on the left) to smallest (on the right).



2. Complete the following metric conversions:

- a) 75 g = 0.075 kg      e) 4.2 kL = 420 000 cL  
 b) 4530 cm = 0.0453 km      f) 83.2 cm = 832 mm  
 c) 0.8 A = 800 mA      g) 975 mg = 0.975 g  
 d) 3200 mA = 3.2 A      h) 975 mg = 0.000975 kg

3. Complete the following table using your knowledge of the Particle Theory of Matter:

STATE of MATTER	FORCES OF ATTRACTION	SPACING OF MOLECULES	MOVEMENT OF MOLECULES
solid	<b>strongest</b>	<b>close together</b>	<b>least</b>
liquid	<b>weaker</b>	<b>farther apart</b>	<b>more</b>
gas	<b>weakest</b>	<b>farthest apart</b>	<b>most</b>

4. Changes of state are examples of physical changes.

- a) Complete the following diagram with labels and arrows to show all of the SIX changes of state. 3. **EVAPOURATION** 4. **CONDENSATION**



5. Name FIVE ways in which you know a chemical change has probably taken place.

colour change      light given off  
bubbles form      heat produced  
precipitate forms

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