

# Physical Science EOC Review

10 pages + 10-page answer key

Physical Science

## PHYSICAL SCIENCE PHYSICS REVIEW

Directions: Complete the following review questions. Write your answers in the space provided.

1. Define the following terms: a. Motion: change in position over time. b. Speed: distance traveled per unit of time. c. Velocity: speed in a given direction. d. Acceleration: change in velocity over time.	2. Calculate the speed of an object that travels 100 meters in 20 seconds. $\text{Speed} = \frac{\text{Distance}}{\text{Time}} = \frac{100 \text{ m}}{20 \text{ s}} = 5 \text{ m/s}$
3. Calculate the velocity of an object that travels 100 meters in 20 seconds and ends up 50 meters from the starting point. $\text{Velocity} = \frac{\text{Displacement}}{\text{Time}} = \frac{50 \text{ m}}{20 \text{ s}} = 2.5 \text{ m/s}$	4. Calculate the acceleration of an object that starts at 0 m/s and reaches 10 m/s in 5 seconds. $\text{Acceleration} = \frac{\text{Change in Velocity}}{\text{Time}} = \frac{10 \text{ m/s} - 0 \text{ m/s}}{5 \text{ s}} = 2 \text{ m/s}^2$

## Newton's Laws of Motion Review

1. Newton's First Law: An object at rest stays at rest and an object in motion stays in motion with the same speed and in the same direction unless acted upon by an unbalanced force.	2. Newton's Second Law: The acceleration of an object is directly proportional to the net force acting on it and inversely proportional to its mass. $a = \frac{F}{m}$	3. Newton's Third Law: For every action, there is an equal and opposite reaction.
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## PHYSICAL SCIENCE CHEMISTRY REVIEW

Directions: Complete the following review questions. Write your answers in the space provided.

1. Define the following terms: a. Matter: anything that has mass and takes up space. b. Element: a substance that cannot be broken down into simpler substances. c. Compound: a substance formed by the chemical combination of two or more elements in a fixed ratio.	2. Calculate the mass of an object that has a volume of 100 cm <sup>3</sup> and a density of 2 g/cm <sup>3</sup> . $\text{Mass} = \text{Density} \times \text{Volume} = 2 \text{ g/cm}^3 \times 100 \text{ cm}^3 = 200 \text{ g}$
3. Calculate the volume of an object that has a mass of 100 g and a density of 2 g/cm <sup>3</sup> . $\text{Volume} = \frac{\text{Mass}}{\text{Density}} = \frac{100 \text{ g}}{2 \text{ g/cm}^3} = 50 \text{ cm}^3$	4. Calculate the density of an object that has a mass of 100 g and a volume of 50 cm <sup>3</sup> . $\text{Density} = \frac{\text{Mass}}{\text{Volume}} = \frac{100 \text{ g}}{50 \text{ cm}^3} = 2 \text{ g/cm}^3$

## Atomic Structure Review

1. Atomic Structure: An atom is made up of three subatomic particles: protons, neutrons, and electrons. Protons are positively charged, neutrons are neutral, and electrons are negatively charged.	2. Periodic Table: A table that organizes elements based on their chemical and physical properties.
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KEY

KEY

ANSWER KEY INCLUDED



1. Calculate the area of a circle with a radius of 5 cm. $\text{Area} = \pi r^2 = \pi (5 \text{ cm})^2 = 25\pi \text{ cm}^2$	2. Calculate the circumference of a circle with a radius of 5 cm. $\text{Circumference} = 2\pi r = 2\pi (5 \text{ cm}) = 10\pi \text{ cm}$
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