

Chapter 5 Study Guide- Physical Science
Test Monday, April 28th, 2014

1. What is a chemical bond? **force of attraction that holds elements together in order to form a unit**
2. Define chemical structure. **the arrangement of atoms in a substance.**
3. What is an ionic bond? **a chemical bond in which one atom loses an electron to form a positive ion and the other atom gains an electron to form a negative ion**
4. An ionic bond is between a **METAL** and a **NONMETAL** .
5. What is a covalent bond? **a chemical bond that involves sharing a pair of electrons between atoms in a molecule**
6. A covalent bond is between a **nonmetal** and a **nonmetal** .
7. What is a metallic bond? **a bond formed by the attraction between positively charged metal ions and the electrons around them.**
8. What are the polyatomic ions? **an ion made of two or more atoms.**
9. What is an ION? **a charged atom**
10. What is a CATION? **A positively charged atom**
11. What is an ANION? **A negatively charged atom**
12. Which part of an ionic compound should be written first? **Metal**

Name the following ionic compounds:

13. FeI₂- **Iron Iodide**
14. MnF₃- **Manganese Fluoride**
15. CrCl₂- **Chromium Chloride**
16. CuS- **Copper Sulfide**

Name the following covalent compounds:

17. As₂O₅- **diarsenic pentaoxide**
18. SiI₄- **silicon tetraiodide**
19. P₄S₃- **tertraphosphorus trisulfide**
20. P₄O₁₀- **tetraphosphorus decoxide**

Write the equation for the following ionic compounds:

21. Lithium Oxide- **Li₂O**
22. Magnesium Nitride- **Mg₃N₂**
23. Sodium Phosphide- **Na₃P**
24. Calcium Bromide -**CaBr₂**

Define/Explain the following

25. What is a subscript? **number written below, tells you how many of each element is present in compound/molecule**
26. What is a superscript? **number written above the element symbol, this is how the oxidation number (charge) is displayed.**

Physical Science Chemical Bond Study Guide

Clemens Wendtner



Physical Science Chemical Bond Study Guide:

Physical Science Chemical Bond Study Guide Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Physical Science Chemical Bond Study Guide**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence.

Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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