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	1.2.4	The molecule is a tetrahedron <sup>™</sup> Die molekule is 'n tetraëder <sup>™</sup> 1 mark/punt	LO2
	1.2.5	Name trigonal pyramid  Noem trigonale piramidale  3 marks/punte	LO4
1.3	1.3.1	The electron pair is equally shared between the two CI atoms and the resulting Cl <sub>2</sub> molecule has no degree of polarity. Die elektronpaar word ewe veel gedeel tussen die twee CI-atome en die resulterende Cl2-molekule het geen mate van polariteit nie. 2 marks/punte	LO2
	1.3.2	The tendency of an atom to pull the bonded pair of electrons towards itself. of Die neiging of van 'n atoom om die gebinde elektronpare na homself te trek of a marks/punte	LO2
	1.3.3	Electronegativity of Na = 6 Electronegativity of Cl = 3  Difference is 3   With a difference of greater than 2.1 electron transfer will take place and the bond elis ionic. (value of 2.1 does not need to be given for this mark) Elektronegatiwiteit van Na = 6 Elektronegatiwiteit van Cl = 3  Verskil is 3   Met 'n verskil van meer as 2,1 sal elektronoordrag plaasvind en die binding is ionies. (waarde van 2.1 hoef nie vir hierdie punt gegee te word nie) 4 marks/punte	LO2

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