



NATIONAL
EXAM CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P1
(CHEMISTRY),
EXEMPLAR

MARKS
100



The question paper consists of 10 pages, including this cover page.
There are 10 questions in this question paper.

QUESTION 3

PHYSICAL SCIENCES PAPER 2 INTERMOLECULAR FORCES

GRADE 11

NOVEMBER 2020

QUESTION 3 MARKS AVAILABLE 10

QUESTION 3 TIME ALLOWED 30 minutes

- 3.1 The following table shows the interaction energy observed during an intermolecular force experiment for halogen molecules in group 17. Complete the table and calculate the mean interaction energy of the four molecules.

Molecule	Interaction energy (kJ/mol)	Percentage (%)
F_2	-100	50
Cl_2	-100	50

- 3.2 Calculate the interaction energy between two molecules of iodine (I_2) and chlorine (Cl_2) at 20 °C if the polar radius of iodine is 1.70 Å.
- 3.3 State what is meant by the term "dipole-dipole interaction".
- 3.4 Explain how the interaction energy of a molecule is affected when it is exposed to air at 20 °C.
- 3.5 Calculate the interaction energy between two molecules of chlorine (Cl_2), each about the same size, from a density gradient tube that contains bromine (Br_2) at the top or bottom layer.
- 3.6 Explain why the bonding angle of chlorine (Cl_2) is larger than that of bromine (Br_2) and the effect of this on the interaction energy of the two molecules.
- 3.7 Calculate the interaction energy of two molecules of chlorine (Cl_2) at 20 °C if their polar radii are different to those in the dipole moment diagram.



- 3.8 Calculate the interaction energy between two molecules of chlorine (Cl_2) at 20 °C if their polar radii are different to those in the dipole moment diagram.



Virtual Tutorship

Physical Science Paper 2 Intermolecular Forces

Investigation

S Ashworth

Physical Science Paper 2 Intermolecular Forces Investigation:

Unveiling the Magic of Words: A Report on "**Physical Science Paper 2 Intermolecular Forces Investigation**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Physical Science Paper 2 Intermolecular Forces Investigation**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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