

- 7 Calcium carbonate,  $\text{CaCO}_3$ , and ammonium sulfate,  $(\text{NH}_4)_2\text{SO}_4$ , may be used by farmers to improve crop yields.

- (a) Complete Table 7.1 by writing the names of the three other elements present in ammonium sulfate and the relative numbers of atoms of each in the compound. One element is done for you.

Table 7.1

element	relative number of atoms
sulfur	1
.....	.....
.....	.....
.....	.....

[3]

- (b) Calculate the relative molecular mass of calcium carbonate,  $\text{CaCO}_3$ .

[Relative atomic masses:  $A_r$ : Ca, 40; C, 12; O, 16]

relative atomic mass = ..... [2]

- (c) Many crops grow best in neutral or weakly alkaline soils. Calcium carbonate is added to acidic soils to increase their pH.

- (i) Suggest the pH number of a weakly acidic soil.

..... [1]

- (ii) State the pH number of neutral soil.

..... [1]

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## **Table of Contents Physical Science Paper 2 December 2014 Exam**

1. Understanding the eBook Physical Science Paper 2 December 2014 Exam
  - The Rise of Digital Reading Physical Science Paper 2 December 2014 Exam
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Paper 2 December 2014 Exam
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Paper 2 December 2014 Exam
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Paper 2 December 2014 Exam
  - Personalized Recommendations
  - Physical Science Paper 2 December 2014 Exam User Reviews and Ratings
  - Physical Science Paper 2 December 2014 Exam and Bestseller Lists
5. Accessing Physical Science Paper 2 December 2014 Exam Free and Paid eBooks
  - Physical Science Paper 2 December 2014 Exam Public Domain eBooks
  - Physical Science Paper 2 December 2014 Exam eBook Subscription Services
  - Physical Science Paper 2 December 2014 Exam Budget-Friendly Options

6. Navigating Physical Science Paper 2 December 2014 Exam eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Science Paper 2 December 2014 Exam Compatibility with Devices
  - Physical Science Paper 2 December 2014 Exam Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science Paper 2 December 2014 Exam
  - Highlighting and Note-Taking Physical Science Paper 2 December 2014 Exam
  - Interactive Elements Physical Science Paper 2 December 2014 Exam
8. Staying Engaged with Physical Science Paper 2 December 2014 Exam
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Paper 2 December 2014 Exam
9. Balancing eBooks and Physical Books Physical Science Paper 2 December 2014 Exam
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Paper 2 December 2014 Exam
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Paper 2 December 2014 Exam
  - Setting Reading Goals Physical Science Paper 2 December 2014 Exam
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Paper 2 December 2014 Exam
  - Fact-Checking eBook Content of Physical Science Paper 2 December 2014 Exam
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

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