

Directed Reading for
Content Mastery**Key Terms**
Chemical Bonds

Directions: Match each term in Column I with its description in Column II. Write the letter of the correct term in the space provided.

Column I

- _____ 1. binary compound
- _____ 2. chemically stable
- _____ 3. nonpolar molecule
- _____ 4. ion
- _____ 5. ionic bond
- _____ 6. polar molecule
- _____ 7. oxidation number
- _____ 8. chemical formula
- _____ 9. covalent bond
- _____ 10. hydrate
- _____ 11. chemical bond
- _____ 12. polyatomic ion

Column II

- a. number that indicates how many electrons an atom must gain, lose, or share to become stable
- b. shorthand that tells what elements a compound contains and the exact number of atoms of each element in a unit of the compound
- c. positively or negatively charged, covalently bonded group of atoms
- d. compound composed of two elements
- e. describes an atom that has a full outermost energy level
- f. molecule that has a slightly positive end and a slightly negative end
- g. the attraction that forms between atoms when they share electrons
- h. the force that holds atoms together in a compound
- i. a compound that has water chemically attached to it
- j. the force of attraction between the opposite charges of the ions in an ionic compound
- k. molecule made of two identical atoms that share the electrons equally
- l. a charged particle that has either more or fewer electrons than protons

Physical Science Chapter 21 Section 21 3 Worksheet Answers

Ralph M. Feather



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Winston,Holt, Rinehart and Winston Staff,2003-03

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Earth Science Ralph M. Feather,1999 Earth science is the study of Earth and space It is the study of such things as the transfer of energy in Earth s atmosphere the evolution of landforms patterns of change that cause weather the scale and structure of stars and the interactions that occur among the water atmosphere and land Earth science in this book is divided into four specific areas of study geology meteorology astronomy and oceanography p 8 9 *Te HS&T J* Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004-02

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Table of Contents Physical Science Chapter 21 Section 21 3 Worksheet Answers

1. Understanding the eBook Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - The Rise of Digital Reading Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - Personalized Recommendations
 - Physical Science Chapter 21 Section 21 3 Worksheet Answers User Reviews and Ratings
 - Physical Science Chapter 21 Section 21 3 Worksheet Answers and Bestseller Lists

5. Accessing Physical Science Chapter 21 Section 21 3 Worksheet Answers Free and Paid eBooks
 - Physical Science Chapter 21 Section 21 3 Worksheet Answers Public Domain eBooks
 - Physical Science Chapter 21 Section 21 3 Worksheet Answers eBook Subscription Services
 - Physical Science Chapter 21 Section 21 3 Worksheet Answers Budget-Friendly Options
6. Navigating Physical Science Chapter 21 Section 21 3 Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Chapter 21 Section 21 3 Worksheet Answers Compatibility with Devices
 - Physical Science Chapter 21 Section 21 3 Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - Highlighting and Note-Taking Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - Interactive Elements Physical Science Chapter 21 Section 21 3 Worksheet Answers
8. Staying Engaged with Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Chapter 21 Section 21 3 Worksheet Answers
9. Balancing eBooks and Physical Books Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Chapter 21 Section 21 3 Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - Setting Reading Goals Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - Fact-Checking eBook Content of Physical Science Chapter 21 Section 21 3 Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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