

Name: Michaela Huber

Matr.:

Klausur: Physikalische Chemie

Physikalische Chemie: Chapter 10: Outer Thermodynamic Energy and Work

Multiple Choice:

1. The electrochemical potential
a. is defined as zero.
b. is arbitrary.
c. is always positive.
d. is always negative.
2. Energy added to a system
a. results mainly in increased translational energy.
b. is lost from the system.
c. is converted to heat.
d. increases the kinetic energy of the system.
3. When a definite quantity of work is done on an object,
a. mechanical work is done.
b. energy is lost from the system.
c. the object is 100% efficient.
d. it gains energy from whatever does the work.
4. A slow cyclic process (quasi-statically slowly) by
a. compression.
b. expansion.
c. isothermal change.
d. adiabatic change.
5. Compression of a gas results in
a. cooling.
b. heating.
c. isothermal change.
d. expansion.
6. Why does liquid water freeze at 0°C?
a. compression.
b. structural rearrangement.
c. transformation.
d. cooling.
7. Which statement is NOT true about heat?
a. It is always positive.
b. Heat always flows from high to low temperature.
c. Heat always flows from high to low temperature.
d. It is not a state function.
8. Which of the following is a thermal coordinate?
a. temperature.
b. heat.
c. work.
d. mass.

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