

Bonding and Reactions

Study Guide

10 pages including KEY

Chemical Bonding and Reactions Study Guide

This study packet will help you study for the Chemical Bonding and Reactions exam by highlighting the most important concepts on the exam. This test will evaluate your understanding of the content you have learned in this course and will be used for the purpose of determining your placement in future courses.

B. VOCABULARY

Atom	Compound	Empirical formula	Molecular formula
Atomic mass	Electronegativity	Hydrocarbon	Hydrocarbon
Atomic number	Electron shell	Hydrogen bond	Hydrogen bond
Chemical bond	Electron configuration	Hydrogen bond	Hydrogen bond
Chemical reaction	Empirical formula	Hydrogen bond	Hydrogen bond

C. READING

1. Why do atoms bond together?

2. What is a covalent bond?

3. What kind of electrons form covalent bonds?

4. What is an ionic bond?

5. What kind of electrons form ionic bonds?

6. Why don't noble gases bond together with other elements?

Write the formula for the following compounds with multiple covalent bonds.

1. Oxygen difluoride _____

2. Oxygen difluoride _____

Write the following compounds with multiple covalent bonds.

3. Water _____

4. Ammonia _____

REACTIONS

What happens in an exothermic reaction?

What happens in an endothermic reaction?

What is an exothermic reaction?

What is an endothermic reaction?

What is the law of conservation of mass?

What is the law of conservation of mass?

A coefficient is a number placed in front of a chemical formula to indicate the number of molecules of that substance that are involved in a chemical reaction.

What is a chemical reaction?

Chemical reactions are changes in the chemical composition of a substance. They are indicated by the formation of new substances from the reactants.

What happens when the temperature of a substance changes?

Substances are substances that are formed by chemical reactions. They are indicated by the formation of new substances from the reactants.

What is a chemical reaction?

A chemical reaction is a process in which one or more substances are converted into one or more new substances.

What are the products in a chemical reaction?

Products are the substances that are formed by chemical reactions. They are indicated by the formation of new substances from the reactants.

What are the products in a chemical reaction?

Products are the substances that are formed by chemical reactions. They are indicated by the formation of new substances from the reactants.

Balance the following reactions:



B. VOCABULARY

Atom - A negatively charged particle that is the basic unit of matter.

Atomic mass - The mass of an atom, measured in atomic mass units (amu).

Atomic number - The number of protons in the nucleus of an atom.

Chemical bond - The force that holds atoms together in a molecule or compound.

Chemical reaction - A process in which one or more substances are converted into one or more new substances.

Compound - A substance formed by the chemical reaction of two or more elements.

Electronegativity - A measure of an atom's ability to attract electrons in a chemical bond.

Empirical formula - The simplest formula for a compound, showing the relative number of atoms of each element.

Molecular formula - The formula for a molecule, showing the actual number of atoms of each element.

Hydrocarbon - A compound consisting of hydrogen and carbon atoms.

Hydrogen bond - A weak attractive force between molecules of water.

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United States Armed Forces Institute

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