

Jan. 2024

The following is an aid to guide your studies in the relevant part of PHY 114. I expect you to read the topics mentioned, and answer the associated questions. On or before 20/2/2024, more questions will be raised on these topics and/or solutions to these questions will be demanded of you by way of homework, and you will be expected to submit them via your class Rep. That will count as ¼ of your CA.

- 1) Study exhaustively chapters on AC circuit and summary of Maxwell's equations.
- 2) Discuss the applications of radioactivity (medical, industrial archaeological, etc). What is binding energy of a nucleus per nucleon? And how do you calculate it?
- 3) Study carefully the contributions of Rutherford and his students to the evolution of modern theory of the atom especially his scattering experiment.
- 4) Study extensively the Bohr's model of the atom and be prepared to show how he used his theory to solve the problem of the Hydrogen spectrum in terms of electronic transitions. What other problems offered a challenge to the physicists of that age?
- 4) The major challenges defying classical physics by the end of 19th century include (list them). How did Einstein solve the problem of photoelectric effect?
- 5) An alpha particle is shot with a speed of 2×10^7 m/s directly towards the nucleus of a gold atom. What is the distance of closest approach to the nucleus? (Take the mass of alpha particle = 6.64×10^{-27} kg, charge on the alpha particle = 3.2×10^{-19} C, atomic number of Au = 79, while a proton charge = $+1.6022 \times 10^{-19}$ C)
- 6) If n stands for integer, and L is the magnitude of angular momentum, and $\hbar$ is the reduced Planck's constant, which of the following is a consequence of Bohr's atomic theory? A) $L = n\hbar$ B) $L = 1/2n\hbar$ C) $L = nh$ D) $L = (n+1)\pi\hbar$
- 7) Calculate the radius of the third excited state in a Bohr atom whose ground state radius is a_0 , and proton number 1. A) $5a_0$ B) $16a_0$ C) $9a_0$ D) $64a_0$
- 8) If the total energy of a Hydrogen-like atom is Y eV. What is the expression for the potential energy of the system in terms of Y ? A) $-Y/2$ eV B) $-2Y$ eV C) $2Y$ eV D) $-Y$ eV
- 9) In a simple model of Bohr Hydrogen atom, it was found that the magnitude of the Coulomb force of attraction between the nucleus and a single electron is given by $|F| = Ke^2/r^2$, which of the following is an expression for the total energy of the Bohr Hydrogen atom assuming that the potential energy of the system is zero at infinity?
- 10) Light of wavelength 430nm is incident on the following metals, which are listed together with their respective work functions (eV). Which metals will show photoelectric emission? I) Cs (2.14 eV) II) K (2.30 eV) III) Na (2.75 eV) IV) Ca (3.20 eV) V) Mo (4.17 eV) ($h = 6.63 \times 10^{-34}$ Js, $c = 3 \times 10^8$ ms⁻¹, 1eV = 1.602×10^{-19} J; all symbols have their usual meaning.)
- 11) A proton and an electron have the same kinetic energy. If the mass of the proton is 1800 times the mass of the electron, find the ratio of their de Broglie wavelengths.
- 12) In the photoelectric experiment performed by Lenard Philip in 1886, which of the following was not an accurate observation? A) intensity of some incident radiation increased the value of the ammeter reading. B) only certain frequencies was able to cause photoelectric emission C) photoelectric emission was instantaneous D) the ammeter read zero when radiations with frequencies equal to or greater than the fundamental frequencies of the material were applied at zero potential difference.
- 13) If the experimental mass of the hydrogen isotope, deuterium is 2.014102u, what is the binding energy per nucleon of Deuterium? (Take mass of a proton = 1.007825u, mass of a neutron = 1.008665u, 1 amu = 931.49 MeV, 1eV = 1.6×10^{-19} J)

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