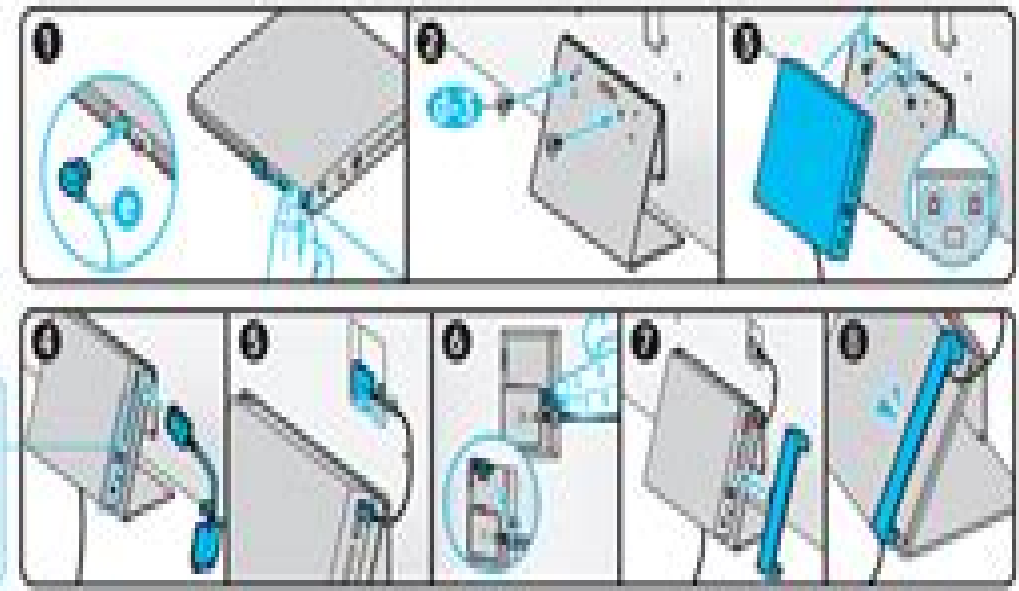
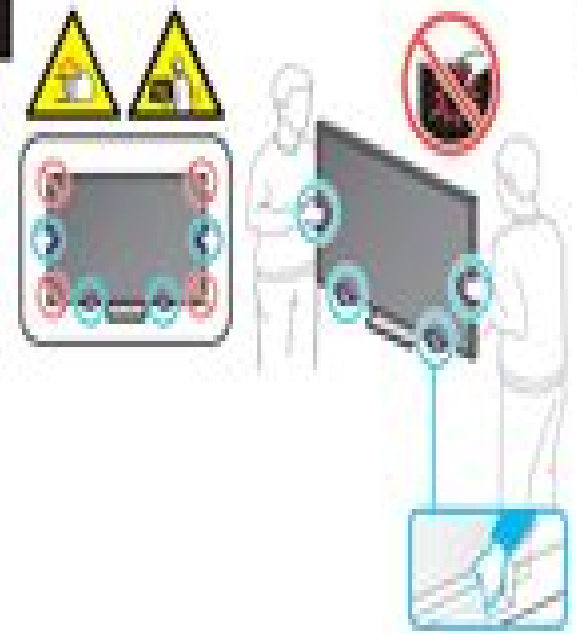


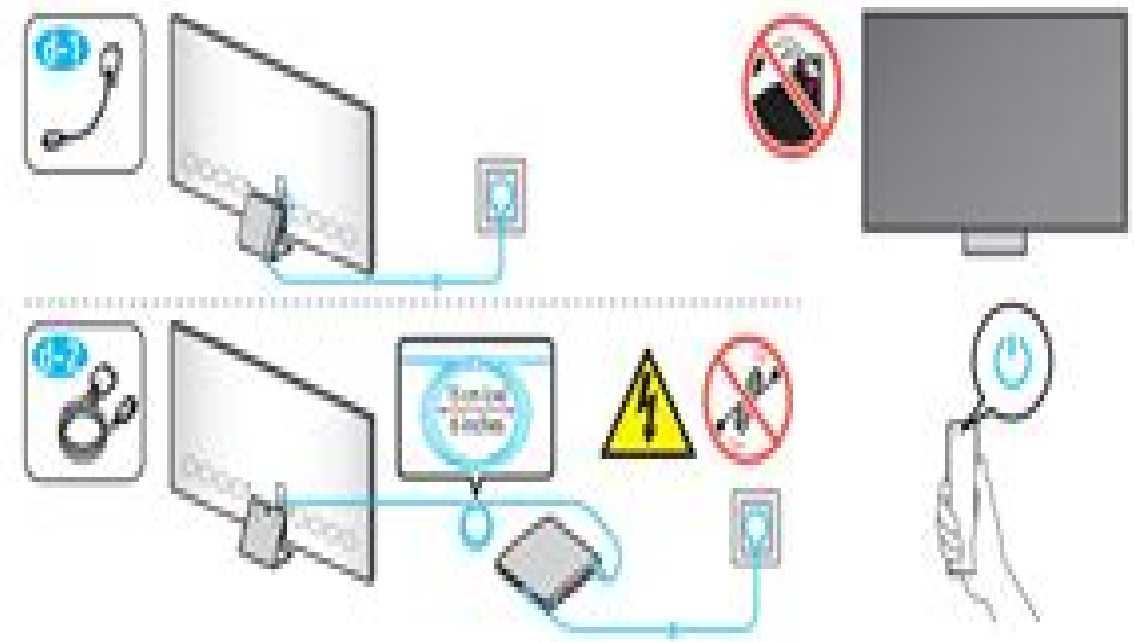
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1 900 kJ mol i 2 1757 kJ mol which of the following identifies the most probable value for the third ionization energy for be  
and provides the best justification

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answer in terms of the number of electron shells 3 on the representation of the periodic table below

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