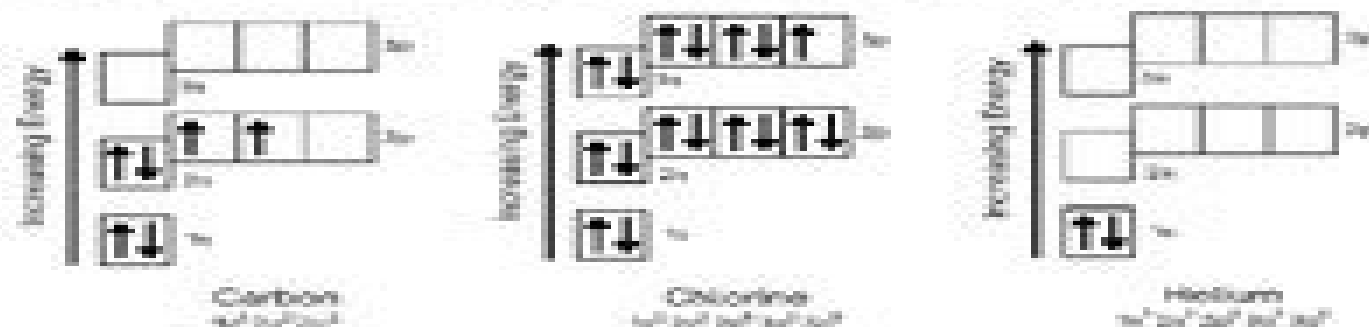


Model II: Electron Configuration with Orbital Diagrams



162 In orbital diagrams, what symbols do you think indicate the following?

- Electrons $\rightarrow$ arrows
- Orbitals $\rightarrow$ boxes
- Sub-level $\rightarrow$ boxes that connect

163 Based on the diagram above, how many electrons does chlorine have? How can you tell?

17, number of arrows

164 When will an electron enter the $2p$ orbital?

After the $2s$ sublevel is filled

165 How do orbital diagrams show the Aufbau principle?

No sublevel receives electrons unless the ones below it are already full.

166 Speculate with your group about why the electrons in carbon's $2p$ sub-level might arrange themselves as depicted above?

e^- have $\ominus$ charge & would rather spread apart if possible

Pogil Hybrid Orbitals

Carl Johan Ballhausen, Harry B. Gray



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