

Final Activity Series

(After 1 week from first activity postcard)

Notes:

After 1 week, students completed activity 1 and they demonstrated a chemical reaction. They reported that silver ions reacted with copper. I asked a student what happened to the silver ions. He said they got into the solution. Which means the silver ions were oxidized. (Which was exactly what we were trying to demonstrate.) We will explore this discussion.

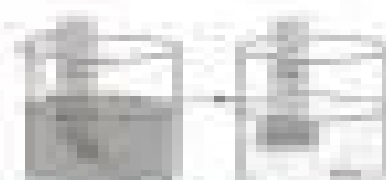
Model 1 – Reaction of Metals



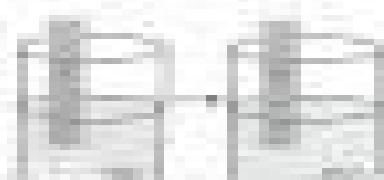
1. $M \rightarrow M^{+} + e^{-}$



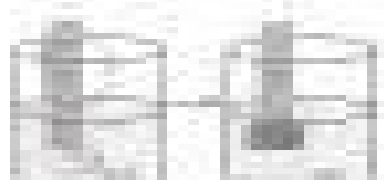
2. $M^{+} + e^{-} \rightarrow M$



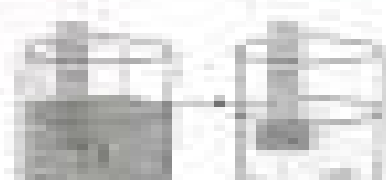
3. $M \rightarrow M^{+} + e^{-}$



4.

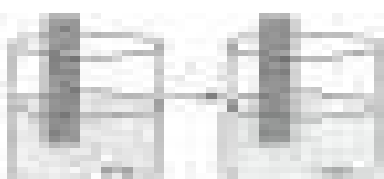


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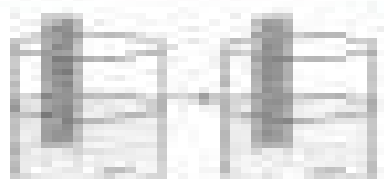


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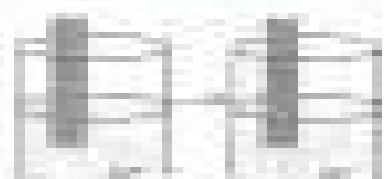
7. $M \rightarrow M^{+} + e^{-}$



8.



9.



10.

After 1 week, students reported evidence for a chemical reaction. They demonstrated a reaction.

The student said:

Pogil The Activity Series Answers

WJ Hussar



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