#### Acids and Bases

How do acids and bases behave in water?

# Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

#### Model 1 - Arrhenius Acids and Bases

Common Name for Aqueous Solution	Chemical Formula	Found in	Tastes	Turns Litmus Paper	Conducts Electricity?	Acid or Base
Acetic acid	HC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> (aq)	Vinegar	Sour	Red	Yes	Acid
Benzoic acid	HC,H,O,(aq)	Food preservative	Sour	Red	Yes	Acid
Phosphoric acid	H,PO <sub>4</sub> (aq)	Soda pop	Sour	Red	Yes	Acid
Hydrochloric acid	HCl(aq)	Stomach acid	Sour	Red	Yes	Acid
Citric acid	H,C,H,O,(aq)	Citrus fruits	Sour	Red	Yes	Acid
Ascorbic acid	H <sub>2</sub> C <sub>6</sub> H <sub>6</sub> O <sub>6</sub> (asq)	Vitamin C	Sour	Red	Yes	Acid
Magnesium hydroxide	Mg(OH),(aq)	Milk of magnesia	Birner	Blue	Yes	Base
Aluminum hydroxide	Al(OH) <sub>s</sub> (aq)	Antacids	Bieter	Blue	Yes	Base
Barium hydroxide	Ba(OH),(aq)	Lubricants	POISON	Blue	Yes	Base
Sodium hydroxide	NaOH(aq)	Drain cleaner	POISON	Blue	Yes	Base

- 1. Refer to Model 1.
  - a. What is the common chemical name for Vitamin C?
  - & Is Vitamin C classified as an acid or a base?
- Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.
- Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?
- Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.

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Joacim Rocklöv

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