# Periodic Trends Practice Worksheet

Use the periodic table and your knowledge of the periodic trends to answer the following questions.

Identify each element as a metal, metalloid, or nonmetal

a germanium \_\_\_\_\_

© zinc \_\_\_\_

phosphorous \_\_\_\_\_\_

(a) lithium

Which of the two species is larger in each of the following pairs?

(a) N<sup>3</sup> or F \_\_\_\_\_ (b) Mg<sup>2</sup> or Ca<sup>2</sup> \_\_\_\_\_ (c) Fe<sup>2</sup> or Fe<sup>3</sup> \_\_\_\_\_

(3) Which of the two species is smaller in each of the following pairs?

(a) K" or Li" (b) Au" or Au3" (c) P3 or N3" \_\_\_\_\_

(i) What atom has the LARGEST atomic radius?

What atom has the SMALLEST atomic radius?

(6) Which is larger?

(a) Mg or Mg<sup>2</sup>\* \_\_\_\_ (b) I or I \_\_\_\_ (c) Cl or Br \_\_\_\_

(d) F or Ne \_\_\_\_ (e) N<sup>3</sup> or P<sup>3</sup> \_\_\_\_

(7) Put the following in order of increasing radius: Br Se2 Sr2 Kr Rb

(8) Circle the correct answer.

(a) Bromine has a higher / lower ionization energy than Lead

(b) Barium has a higher / lower ionization energy than Radon

© Fluorine has a higher / lower ionization energy than Arsenic

(3) Lithium has a higher / lower ionization energy than Cesium

Which of the two species is larger in each of the following pairs?

(a) Mg or Sr

(b) Be or N (c) P or Cl (d) Ni or Br

Arrange the following in order of increasing electronegativity: S, O, Ne, Al

# **Periodic Trends Practice Problems With Answers**

# **A Gutmann**

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