

5.1 Atomic structure and the Periodic Table

You will learn:

- To understand how the position of an element in the Periodic Table can be used to predict its atomic structure and chemical and physical properties.
- To understand that models and analogies can change according to scientific evidence
- To describe the uses, strengths and limitations of models and analogies
- To represent scientific ideas using recognised symbols or formulae

1.

Draw a line to match the word with its correct meaning.

The first line has been drawn for you.

Word	Meaning
Electrons	Particles with no charge found in an atom
Neutrons	The central part of the atom
Protons	The paths or orbits that electrons move along in the atom
Nucleus	Very small negatively charged particles in an atom
Shells	Particles with a positive charge found in an atom

2.

Complete the sentences.

Use the words in the box.

atomic	mass	protons	electrons	nucleus
	shells	neutrons	atom	

- a An _____ number is the protons an _____ of an element has.
- b A _____ number is the total number of _____ and _____ found in the nucleus of an atom.

Periodic Table Workbook Answers Chemistry

Brian Arnold



Periodic Table Workbook Answers Chemistry:

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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

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