

Mole Ratios

How can the coefficients in a chemical equation be interpreted?

Why?

A balanced chemical equation can tell us the number of reactant and product particles (ions, atoms, molecules or formula units) that are necessary to conserve mass during a chemical reaction. Typically when we balance the chemical equation we think in terms of individual particles. However, in real life the reaction represented by an equation occurs an unimaginable number of times. Short of writing very large numbers (10^{23} or larger) in front of each chemical in the equation, how can we interpret chemical equations so that they more realistically represent what is happening in real life? In this activity you will explore the different ways a chemical reaction can be interpreted.

Model 1 – A Chemical Reaction



1. Consider the reaction in Model 1.
 - a. What are the coefficients for each of the following substances in the reaction?



- b. Draw particle models below to illustrate the reaction in Model 1.



2. Consider each situation below as it relates to the reaction in Model 1.
 - a. Calculate the amount of reactants consumed and products made.
 - b. Record the ratio of N_2 to H_2 to NH_3 . Reduce the ratio to the lowest whole numbers possible.

	N_2 Consumed	H_2 Consumed	NH_3 Produced	Ratio $\text{N}_2:\text{H}_2:\text{NH}_3$ (reduced)
For a single reaction, how many molecules of each substance would be consumed or produced?	1 molecule	3 molecules	2 molecules	1:3:2
If the reaction occurred one hundred times, how many molecules would be consumed or produced?	100 molecules	300 molecules	200 molecules	1:3:2
If the reaction occurred 538 times, how many molecules would be consumed or produced?	538 molecules	1614 molecules	1076 molecules	1:3:2

Pogil Chemistry Mole Ratios Answers Model 2

L Darling-Hammond



Pogil Chemistry Mole Ratios Answers Model 2:

Unveiling the Power of Verbal Beauty: An Psychological Sojourn through **Pogil Chemistry Mole Ratios Answers Model 2**

In a world inundated with screens and the cacophony of instant conversation, the profound power and psychological resonance of verbal artistry frequently diminish in to obscurity, eclipsed by the constant onslaught of noise and distractions. Yet, set within the musical pages of **Pogil Chemistry Mole Ratios Answers Model 2**, a interesting perform of fictional splendor that pulses with organic feelings, lies an wonderful trip waiting to be embarked upon. Composed with a virtuoso wordsmith, this exciting opus guides viewers on a psychological odyssey, delicately revealing the latent potential and profound impact stuck within the complicated internet of language. Within the heart-wrenching expanse with this evocative evaluation, we can embark upon an introspective exploration of the book is key subjects, dissect its interesting writing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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