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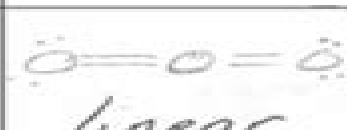
## Molecular Geometry

How can molecular shapes be predicted using the VSEPR theory?

### Why?

When you draw a Lewis structure for a molecule on paper, you are making a two-dimensional representation of the atoms. In reality however, molecules are not flat—they are *three-dimensional*. The true shape of a molecule is important because it determines many physical and chemical properties for the substance. In this activity you will learn how to predict molecular shapes.

### Model 1 – Lewis Structures

| Lewis Structures         |  | 3-D Molecular Shape                                                                   |
|--------------------------|--|---------------------------------------------------------------------------------------|
| 1. $\text{H}_2\text{CO}$ |  |    |
| 2. $\text{BeF}_2$        |  |    |
| 3. $\text{CH}_4$         |  |    |
| 4. $\text{NH}_3$         |  |    |
| 5. $\text{H}_2\text{O}$  |  |   |
| 6. $\text{CO}_2$         |  |  |

Lone pair = ..

# Pogil Activities For High School Molecular Geometries Answers

**O García**



## **Pogil Activities For High School Molecular Geometries Answers:**

## **Pogil Activities For High School Molecular Geometries Answers** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Pogil Activities For High School Molecular Geometries Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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