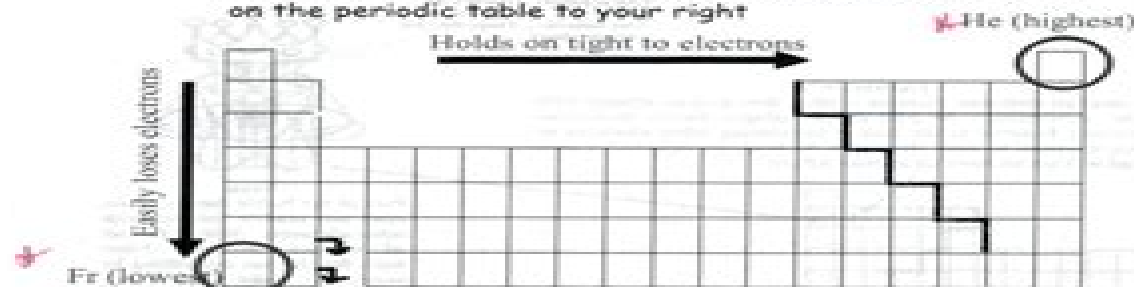


b. Why do elements get larger as you go down a group? Additional energy levels are added to the electron cloud as you go down a group.

Ionization Energy:

1. In simple terms, explain ionization energy. The energy required to remove an electron from a neutral atom.
2. Using the information in figure 5-15 (Pg 143) in your book, write the ionization energy for alkali metals & period #2 on the periodic table to your right



3. On the arrows on the diagram, indicate the trend for ionization energy as you move down a column and across a row. (Options: easily loses an electron OR holds on tight to electrons)
4. Of all the elements on the periodic table, identify & circle the element with the highest & lowest ionization energy.

5. Rank the following elements from highest to lowest ionization energy:

a) Be, Ba, Mg = Be, Mg, Ba

b) Li, F, B = F, B, Li

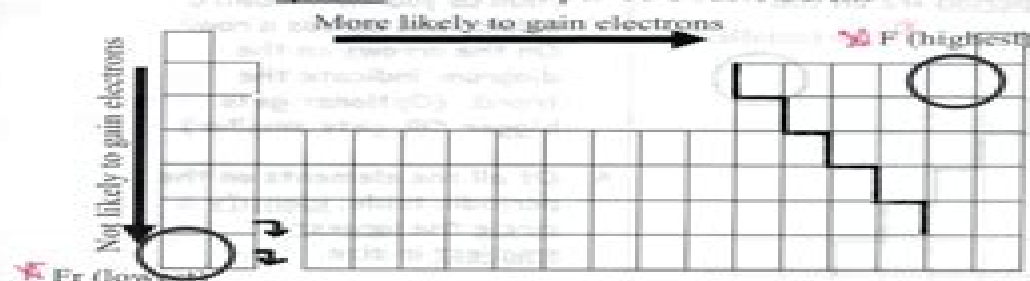
6. Why does it take less energy to remove an electron from a large atom compared to a small atom? Their electrons are closer to the nucleus, so the protons have a larger "pull" on them.

It takes less energy to remove an electron from a large atom because

Electronegativity: the effective nuclear charge decreases and electrons are not held as tightly since

1. Explain Electronegativity: The ability of an atom to get electrons from other atoms.

2. Using the information in figure 5-20 (pg 151) in your book, write the electronegativity for the Halogens and period #2 on the periodic table below.



3. On the arrows on the diagram, indicate the trend for electronegativity as you move down a column and across a row. (Options: more likely to gain electrons, not likely to gain electrons)

4. Of all the elements on the periodic table, identify & circle the element with the highest & lowest electronegativity.

5. Rank the following elements from most to least electronegative element:

a) Be, Ba, Mg = Be, Mg, Ba

b) Li, F, B = F, B, Li

6. Why are small atoms better at pulling electrons off of larger atoms than vice versa?

Their electrons are closer to the nucleus, so the protons have a larger "pull" on the electrons in other atoms.

not detailed enough
The increased number of energy level(s) increases the distance over which nucleus must pull and therefore reduces the attraction for electrons } *AP answer*

Pogil Cracking The Periodic Table Answer

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Pogil Cracking The Periodic Table Answer:

Atoms, Molecules & Elements: Patterns In the Periodic Table Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice Patterns In the Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives

Elements and the Periodic Table, Grades 5 - 12 Theodore A. Abbigy, 2001-01-01 This informative classroom supplement is a great introduction to the periodic table explored in sequential form It includes activities transparency masters a teacher's guide an element game quizzes tests rubrics and answer keys Unit topics include discovering what elements are the uses of the elements element symbols periodic table organization and more Mark Twain Media Publishing Company specializes in providing captivating supplemental books and decorative resources to complement middle and upper grade classrooms Designed by leading educators the product line covers a range of subjects including mathematics sciences language arts social studies history government fine arts and character Mark Twain Media also provides innovative classroom solutions for bulletin boards and interactive whiteboards Since 1977 Mark Twain Media has remained a reliable source for a wide variety of engaging classroom resources

Elements and the Periodic Table, Grades 5 - 12 Theodore S. Abbigy, 2013-01-02 Aligned to Common Core State Standards Elements and the Periodic Table present the basics of the Periodic Table in an easy to understand easy to master way It contains fun activities transparency masters quizzes tests rubrics grading sheets and more From basic elements to table organization Elements and the Periodic Table is the essential handbook for middle school science

The Basics of the Periodic Table Leon Gray, 2013-12-15 A sweeping history of both the discovery and classification of elements and the development of the modern periodic table Included are discussions of the discovery of matter atoms atomic structure molecules compounds ions and isotopes as well as the first identifications of the 118 and counting elements and the various ways they have been classified and organized by prominent scientists up to the present day periodic table Instruction in how to read the periodic table is accompanied by examinations of the various groups of elements their location on the table and their properties and practical uses This text strongly supports Common Core Standards for the reading of scientific and technical texts and accounts and furnishes ample opportunities to summarize cite evidence and analyze connections between ideas individuals and events

Atoms,

Molecules & Elements: The Periodic Table Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice The Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives

Cracking the Elements Rebecca Mileham, 2018-09-06 From the earliest known elements to those named in 2016 this book takes a comprehensive look at the development of the periodic table and reveals untold stories unsung pioneers and plenty of fascinating science along the way In twelve illustrated chapters the book makes sense of the patterns and groups within the periodic table introducing each of the 118 known elements individually and exploring questions including Why did the history of fizzy water give early chemistry a sparkle How did hydrogen reveal the structure of the atom What was the Bunsen burner s role in discovering new elements Which of the alkaline earth metals accounts for a kilogramme of your weight Why is Marie Curie such a scientific star How do tungsten and vanadium explain the secret of super sharp Syrian swords Who discovered the most elements in the periodic table What made nihonium element 113 such a wonderful new year s gift for Japan Is glass a liquid or a solid How did nitrogen fulfill the alchemists dream Would you have smeared antimony on your face if you d lived in ancient Egypt Why might naked mole rats have clues for surviving a heart attack How did the Haya people of Tanzania make steel 1500 years ago What makes xenon a great anaesthetic and why can t all patients use it Might there be a pattern in yet undiscovered elements beyond number 118

Leveled Texts: The Periodic Table Joshua BishopRoby, 2014-01-01 All students can learn about the periodic table through text written at four different reading levels Symbols on the pages represent reading level ranges to help differentiate instruction Provided comprehension questions complement the text

Quiz Yourself Clever! The Periodic Table DK, 2024-10-03 Get to grips with the periodic table with this fact packed quiz book Learn all the ingredients that make up the universe in this quiz book for children aged 9 that breaks down the periodic table Quiz Yourself Clever The Periodic Table goes through all 118 elements in an accessible and kid friendly way showing the pure form of every element through an eye catching image surrounded by quiz style questions This periodic table quiz book for children offers Fast and fun learning through the use of detailed images and interesting facts Hundreds of quick fire questions for children to quiz themselves and challenge friends and family Information that has been endorsed and authenticated by experts Vivid pictures and fast facts will allow children to gather a wealth of knowledge in a fun way Turn

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The Periodic Table Adrian Dingle, Dan Green, 2014 The original Basher Science made even better

The Elements Book DK, 2017-04-04 Richly illustrated with over a thousand photos and dazzling details of the elements that make up the physical world Written in association with the renowned Smithsonian Institution Does your little chemist have questions about the stuff that everything is made of This visual reference book covers each of the 118 elements and includes a glossy pull out poster of the periodic table This encyclopedia is a superb introduction to the subject of chemistry Written with kids ages 9 to 12 in mind using easy to understand language and straightforward fun facts There s information on the scientists that made the first discoveries and spectacular photos of large natural features along with a simple explanation of what an atom is Find out which of the things we see every day contain these common and unusual elements There s so much to discover about different elements Explore their atomic structure with the number of electrons protons and neutrons and the three states of solid liquid or gas Kids will learn that the copper used in computer motherboards is also what the Statue of Liberty is made of and why it s green Also learn about elements like zinc why Japan s Akashi Bridge is coated in zinc and why zinc is used in the soles of boots to make the rubber tougher Each element is shown in its pure form in a stunning series of photos that will keep children engrossed in elemental science The poster included with this education book is an added learning tool that shows how the elements are arranged on the periodic table It s easier than ever to look up the basics of chemistry From Ac to Xe and all the elements between The multitude of photos in this appealing format makes learning the fundamentals of chemistry simple and enjoyable This visual reference guide provides the reader with an overview of the most fascinating facts about the elements within us and around us Concise and bite sized information makes it easy for young scholars to follow Eye catching and captivating photos of raw elements and what they are used in

New Ideas in Chemistry from Fresh Energy for the Periodic Law Henry Bent, 2006-09-05 New Ideas calls to mind Aristotle s synopsis of the Iliad and the Odyssey Woman abducted Long war One guy has a hard time getting home End of story The rest is episodes Similarly here Chemical capture of the Left Step Periodic Table One element finds a new home The noblest of the noble gases is not a Noble Gas End of story The rest is novel consequences of the Noble Gas Conclusion Among them overlooked Rules of Triads Block Sizes and Full Shells overlooked block to block trends and a correspondence between elements ordinal numbers in their Groups and orbital s radial quantum numbers and recognition that Pauli s explanation of Periodicity s magic numbers 2 8 18 got the right answer the Pauli Exclusion Principle for the wrong reason New Ideas ends with suggestions for streamlining the teaching of

the mole concept chemical bonding and thermodynamics in order to provide room in the chemistry curriculum for a more thorough treatment of Periodic System Systematics

The Periodic Table Sharon Katz Cooper, 2007 An introduction to chemistry's periodic table of elements *Atoms, Molecules & Elements Gr. 5-8* George Graybill, 2007-09-01 Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource makes the periodic table easier to understand Begin by answering what are atoms See how the atomic model is made up of electrons protons and neutrons Find out what a molecule is and how they differ from elements Then move on to compounds Find the elements that make up different compounds Get comfortable with the periodic table by recognizing each element as part of a group Examine how patterns in the period table dictate how those elements react with others Finally explore the three important kinds of elements metals nonmetals and inert gases Aligned to the Next Generation Science Standards and written to Bloom's Taxonomy and STEAM initiatives additional hands on experiments crossword word search comprehension quiz and answer key are also included *Periodic Table* DK, 2020-01-02 A brand new science ebook for kids joins this hugely successful mini encyclopedia series including lot of information Discover all the major elements of the periodic table arranged in chapters according to their group including alkaline earth metals lanthanides and noble gases Every important element from hydrogen via carbon oxygen and gold to oganesson that's element number 118 is presented here Each element is featured in its own catalogue style entry with a photo of the element in its raw state or in use a caption explaining important information ranging from where it's found to its main uses and a fact list presenting the key data including the date of discovery and all atomic information as found on the periodic table Plus there are stunning full page photos showing elements as you have never seen them before perhaps inside machinery that you can't normally delve into or in stunning macro photography that reveals microscopic details invisible to the human eye The style of the Pocket Eyewitness series is perfect for all children from reluctant readers who can easily digest the key points through to budding Marie Curies and Louis Pasteurs who want to know more about the most essential particles on the planet

Mastering the Periodic Table Linda Trombley, Faye Williams, 2000 Whether students are studying chemistry biology or other sciences the periodic table is a vitally important tool These 50 word games puzzles and other creative activities unlock the nature of the various elements while explicating periodicity atomic structure element groups and more Complete teacher support includes background information answer keys and materials lists

Decoding the Periodic Table Jurjen Van Van der Wal, 2010-06-16 Decoding the Periodic Table In the year 1809 Russian Chemist Dimitri Ivanovich Mendeleev published the first Chemical Periodic Table of the Elements Since then scientists developed a theory that every element consists of a tightly packed mass of protons and neutrons which is surrounded by concentric spheres which are occupied by orbiting electrons This concept is known as the Standard Model but it is flawed because it cannot accommodate gravity For many decades researchers including this author in the late 1960's tried to formulate theories that in a comprehensive way would unite all physical designs and phenomena under a common

denominator but to no avail The String Theory is one of them but it has found no success In 1989 the author of the book made a fresh start again with the notion that particles such as the proton neutron and electron would not be round but that they would have straight lined square or cubic configurations because the Periodic Table s had quadratic numbers in their quantities of elements such as Period 1 with 2 elements 2 2x12 Periods 2 3 8 2 x 22 Periods 4 5 18 2 x 32 and Period 6 7 32 2 x 42 This might indicate square structural relationships lets give it a try It worked it took 20 years to do it it was a fascinating journey falling down and getting up again and again working without a computer with all figures hand drawn and discovering that the forces that make the nuclear mass of the neutron do that by multiplying two force filled cubic volumes as in $3F \times 3F \times 3F$ x $3F \times 3F \times 3F$ 1838 2656 **Inspecting Elements & the Periodic Table** Jessica Rusick,2022-08-01 This title provides an overview of elements and the periodic table Text includes a simple overview of elements and examines compounds categories of elements periods groups atomic numbers properties isotopes and allotropes Information is explained using real world examples and supported with graphics and photos This book concludes with two simple kid friendly experiments Aligned to Common Core standards and correlated to state standards Checkerboard Library is an imprint of Abdo Publishing a division of ABDO **Science Spectrum** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004 **Chemical Elements Pocket Guide** Coventry House Publishing,2020-02-08 The Chemical Elements Pocket Guide serves as a portable reference for quick study and efficient review of the 118 elements on the periodic table This on the go resource details the physical and atomic properties of each element as well as their history and characteristics in bullet point format The book s small trim size 4 25 x 6 8 inches is intended to fit inside a lab coat pocket and the bound design means you no longer need to carry loose bulky flashcards that can be misplaced or destroyed Includes the updated names nihonium moscovium tennessine and oganesson for elements 113 115 117 and 118 respectively Information provided includes Atomic number Atomic symbol Element category Standard state Atomic mass Electron configuration Oxidation states Electronegativity Atomic radius Ionization energy Electron affinity Melting point Boiling point Density Year discovered Discovered by Appearance Natural occurrence Interesting fact *Periodic Table 91 Success Secrets - 91 Most Asked Questions on Periodic Table - What You Need to Know* Patrick Stein,2014-10-01 Probably The Best Periodic table Guide To Date There has never been a Periodic table Guide like this It contains 91 answers much more than you can imagine comprehensive answers and extensive details and references with insights that have never before been offered in print Get the information you need fast This all embracing guide offers a thorough view of key knowledge and detailed insight This Guide introduces what you want to know about Periodic table A quick look inside of some of the subjects covered Extended periodic table Predicted properties of undiscovered elements Extended periodic table Elements 165 to 172 Extended periodic table Eka superactinides Extended periodic table 8s elements Transition metal Position in the Periodic Table Block periodic table Relativistic effect Periodic table deviations History of the periodic table Ancient times Period periodic table Period 1

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