

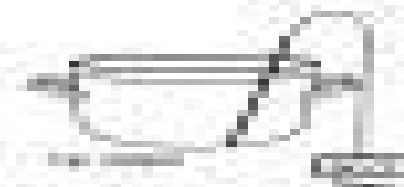
Calorimetry

What is calorimetry? How does it work? What is its purpose?

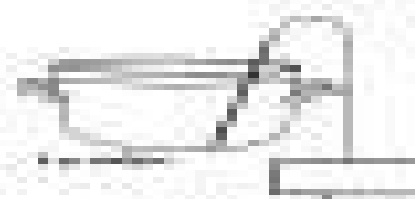
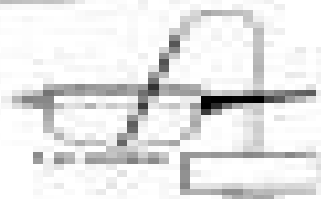
When a substance is heated, the temperature of that substance increases. Will the amount of energy (heat) required to heat different substances to the same temperature increase? Will the same amount of energy be needed to heat different temperatures? Increase or decrease amounts of the same substance to different temperatures will require how much, temperature, heat energy, and the type of substance are related.

Model 1 – A Plot of Mass

Calorimetry



The energy to heat a substance is proportional to mass.



1. In Model 1, which substance has the same amount of mass? Substance A

2. Compare the graphs for the two substances.

a. How much heat energy was added to the substance? 100 J

b. How much heat energy was added to the substance? 100 J

c. In Model 1, which substance has the same amount of mass? Substance A

What is the purpose of calorimetry? To measure the heat energy of a substance.
 What is the purpose of calorimetry? To measure the heat energy of a substance.
 What is the purpose of calorimetry? To measure the heat energy of a substance.

Pogil Activities For High School Chemistry Calorimetry Answer Key

JR Anderson



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