

CLO- Classification of Elements & Periodicity in Properties

MULTIPLE CHOICE QUESTIONS:

Choose the one correct option in the following:

- Identify the least stable ion amongst the following :
(a) Li^+ (b) Be^+ (c) B^+ (d) C^+
- The correct order of radii is :
(a) $\text{N} < \text{Be} < \text{B}$ (b) $\text{F}^- < \text{O}^{2-} < \text{N}^{3-}$ (c) $\text{N} < \text{Li} < \text{K}$ (d) $\text{Fe}^{2+} < \text{Fe}^{3+} < \text{Fe}^{4+}$
- Which one of the following has largest size :
(a) Na (b) Na^+ (c) Na^- (d) Li^+
- The number of elements in the 4th period of periodic table is :
(a) 8 (b) 10 (c) 18 (d) 32
- Which of the following has least electron affinity (electron gain enthalpy) :
(a) O (b) N (c) Ar (d) F
- An element of atomic number 29 belongs to which of the following block of the periodic table :
(a) s-block (b) p-block (c) d-block (d) f-block
- Which one of the following arrangements not truly represents the property indicated against it:
(a) $\text{Br}_2 < \text{Cl}_2 < \text{F}_2$ – electronegativity (b) $\text{Br}_2 < \text{F}_2 < \text{Cl}_2$ – Electron Affinity
(c) $\text{Br}_2 < \text{Cl}_2 < \text{F}_2$ – Bond energy (d) $\text{Br}_2 < \text{Cl}_2 < \text{F}_2$ – Oxidising Power
- Alkali metals in each period have :
(a) Smallest size (b) Lowest IE (c) Highest IE (d) Highest EN
- A sudden large jump between the value of second and third ionization energies of elements would be associated with which of the following electronic configuration :
(a) $2\text{p}^4 3\text{s}^1$ (b) $3\text{s}^2 3\text{p}^1$ (c) $3\text{s}^2 3\text{p}^2$ (d) $2\text{p}^4 3\text{s}^2$
- Which of the following pairs are chemically dissimilar :
(a) Na and K (b) Ba and Sr (c) Zr and Hf (d) Ca and Zn
- Collective name given to the elements with outer shell configuration $ns^2 np^6$ is :
(a) Chalkogens (b) Alkaline earths metals (c) Transition metals (d) Nobel gases
- Modern extended periodic table was given by :
(a) Bohr (b) Lothar mayer (c) Moseley (d) None alone
- The most electronegative elements is :
(a) Hydrogen (b) Fluorine (c) Chlorine (d) Oxygen
- The element having electronic configuration :
 $[\text{Kr}]4\text{d}^{10}4\text{f}^{14}5\text{s}^2 5\text{p}^6 5\text{d}^1 6\text{s}^2$ belongs to
(a) s-block (b) p-block (c) d-block (d) f-block
- An elements R forms the highest oxide R_2O_5 . R belong to :

Periodicity Multiple Choice Questions

Disha Experts

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the 'Disha Experts' text.

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In a global inundated with displays and the cacophony of instantaneous interaction, the profound power and mental resonance of verbal artistry frequently fade in to obscurity, eclipsed by the continuous onslaught of sound and distractions. Yet, nestled within the musical pages of **Periodicity Multiple Choice Questions**, a charming function of fictional splendor that pulses with natural emotions, lies an unforgettable trip waiting to be embarked upon. Penned with a virtuoso wordsmith, this magical opus instructions viewers on a psychological odyssey, softly exposing the latent potential and profound influence embedded within the elaborate web of language. Within the heart-wrenching expanse with this evocative evaluation, we will embark upon an introspective exploration of the book is main styles, dissect its charming publishing fashion, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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