#### **Chemical Periodicity**

Name <u>Mitchell Wade</u> Section Number <u>015</u>

OBJECTIVE: To demonstrate the periodic law.

CONCEPTS: Periodic law, Periodic trends

Intracouction: In the 1860's about 60 elements were known. From their physical properties, the elements were classified as metals, nonmetals, or metalloids. Metals have low electrical resistance (high conductivity), have a metallic luster, and are malleable. Non-metals have high resistance, low luster, and are brittle. Metalloids have properties between those of non-metals and metals. Chemists of the time further classified the elements into groups. The elements in a particular group have similar chemical and/or physical properties. This classification led to the discovery of the periodic table and the periodic law. In the periodic table, elements are assigned to a period and a group—groups are the columns and the periods are the rows of the table. The periodic law states that the properties of the chemical elements vary with the atomic number in a systematic way. We now understand that the properties depend on the electronic structures of the atoms.

#### PROCEDURE:

 Find the station labeled "Halogens: states of matter." Observe the states of matter of the three samples. How does the state of matter change as you go down the column of the periodic table?

Starts as gas and goes to solid

The melting points of the halogen elements that you observed are listed below. Based on the trend that you see, estimate the melting point of Fluorine.

| Element | Melting Point (°C) |
|---------|--------------------|
| F       | -200               |
| CI      | -101.5             |
| Br      | -7.35              |
| 1.      | 113.75             |

Find the station labeled "Metallic Solutions." Fill out the chart below. For the unlabeled mystery elements, you will have to guess which type of metal they are (transition, alkali, or alkaline earth) based on the trends you see.

| Solution<br>Number | Element | Color of Water Solution | Type (transition metal, alkali metal,<br>alkaline earth metal) |
|--------------------|---------|-------------------------|--|
| 1                  | Co      | Red                     | Transition metal   |
| 2                  | 8a      | Clear                   | Alkaline earth metal   |
| 3                  | K       | clear                   | Alkali metal   |
| 4                  | Ni      | green                   | Transition metal   |
| 5                  | Na      | clear                   | Alkali metal   |
| 6                  | Cu      | blue                    | Transition metal   |
| 7                  | Mg      | clear                   | Alkaline earth metal   |

# **Periodicity Lab Report**

**Michal Rosen-Zvi** 

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Scientific and Technical Aerospace Reports, 1994 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical U.S. Government Research Reports ,1964 **Information Database Sustainable Green Chemistry** Mark Anthony Benvenuto, 2017-04-10 Sustainable Green Chemistry the 1st volume of Green Chemical Processing covers several key aspects of modern green processing The scope of this volume goes beyond bio and organic chemistry highlighting the ecological and economic benefits of enhanced sustainability in such diverse fields as petrochemistry metal production and wastewater treatment The authors discuss recent progresses and challenges in the implementation of green chemical processes as well as their transfer from academia to industry and teaching at all levels Selected successes in the greening of established processes and reactions are presented including the use of switchable polarity solvents actinide recovery using ionic liquids and the removal of the ubiquitous bisphenol A molecule from effluent streams by phytodegradation Report of the **Annual Meeting** British Association for the Advancement of Science. Meeting, 1920 **Report of the ... Meeting** British Association for the Advancement of Science. Meeting, 1920 Cloud Nucleation C. Guy Suits, 2013-10-22 The Collected Works of Irving Langmuir Volume 11 Cloud Nucleation presents the papers written by Dr Irving Langmuir on the study of artificial nucleation of clouds to produce precipitation The compilation provides the study of every possible aspect of cloud and weather modification This volume contains chapters that are devoted to the discussion of such subjects as the nucleation of clouds with dry ice the analysis of photographic data showing the effects produced by supercooled clouds with ice nuclei the production of rain by a chain reaction in cumulus clouds at temperatures above freezing nucleation of clouds with silver iodide and basic discoveries underlying weather modification Meteorologists will find the book interesting and invaluable

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