

Step 8. Periodic table trends

Electronegativity

Ionization energy

He

Atomic radius

The periodic table displays various elements with their symbols and atomic numbers. Trends are indicated by colored arrows:

- Electronegativity:** Indicated by a black arrow pointing right across the top.
- Ionization energy:** Indicated by a green arrow pointing right across the top and a green arrow pointing up on the right side.
- Atomic radius:** Indicated by a red arrow pointing down on the left side and a red arrow pointing left across the bottom.
- Non-metallic properties:** Indicated by a purple arrow pointing up and to the right.
- Metallic properties:** Indicated by a blue arrow pointing down and to the right.

Ionization Energy

Electronegativity

Atomic radius

An arrow head shows the INCREASE of the property

Example: comparing K atom with Br atom

- Atomic radius of K is HIGHER than atomic radius of Br
- Ionization energy of K is LOWER than ionisation energy of Br
- Metallic properties of K are HIGHER than metallic properties of Br



Periodic Trends Answer Key Section Review

G Psacharopoulos

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Periodic Trends Answer Key Section Review:

Reviewing **Periodic Trends Answer Key Section Review**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Periodic Trends Answer Key Section Review**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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