

# RS267TDRS

26 cu. ft. Side by Side Refrigerator

SAMSUNG

## Features:

- Large Capacity – 26 cu. ft.
- Energy Star Rated: 502 kWh/year
- External Filtered Water and Ice Dispenser
- LED Tower Lighting
- Twin Cooling System®
- Multi Air Flow
- Six Temperature Sensors
- Power Freeze and Power Cool Options
- Cool Select Zone™ Drawer with Temperature Control
- Tempered Glass Spill Proof Shelves
- Gallon Door Bins

## Convenience:

- Door Alarm
- Filter Indicator
- Wine Rack



Samsung's innovative Side by Side Refrigerator offers 26-cu. ft. of capacity and the ultimate in food preservation with Twin Cooling System® technology.

Ranked "Highest in Customer Satisfaction with Refrigerators, Five Years in a Row" by J.D. Power and Associates.\*

### Design/Style:

- Side by Side Door Configuration
- Concealed Hinges
- External Digital Blue LED Display



### Color:

Stainless Steel  
(shown here)  
White

Black

Stainless Platinum



### Model #:

RS267TDRS

RS267TDVP

RS267TDBP

RS267TDPN

\* Samsung received the highest numerical score for refrigerators in the proprietary J.D. Power and Associates 2006-2007 Major Home Appliance Study, and 2008-2009 Kitchen Appliance Study™. 2007-2008 received award for satisfaction with side-by-side French door refrigerators. Study based on 11,170 total responses measuring 13 brands and measured opinions of consumers who purchased refrigerators from a retail store, their new-home builder during the previous 24 months. Proprietary study results are based on experiences and perceptions of consumers surveyed in March-April 2008. Your experiences may vary. Visit [jdpower.com](http://jdpower.com).

# Samsung Rs267td Refrigerators Owners Manual

**Jin-Ying Zhang**



**Samsung Rs267td Refrigerators Owners Manual:**

**Household Refrigerator** Copeland Refrigeration Corporation,1938\*      **Coldspot Frostless Refrigerator-freezer**  
Sears, Roebuck and Company,1960\*      *Side-by-side Refrigerator Service Manual* Master Care,1982      *The New Frigidaire*  
*Refrigerator* ,1960 Covers the Cyclomatic C1020 De luxe 1130 Master 870 Master 710 De luxe 870 De luxe 710 and Master  
510 models      **Sparton Refrigerator Service Manual** Sparks-Withington company, Jackson, Mich. Service Division,1936\*  
**Gibson Freez'r Shelf Refrigerator** Gibson Electric Refrigerator Corporation,1938      *Coldspot Frostless*  
*Refrigerator-Freezer: Owner's Manual* ,      *Sparton Refrigerator Service Manual* Sparks-Withington Company (Jackson,  
Mich.). Service Division,1936      *Basic Service Manual Top-mount Refrigerators* Roper Sales Corp,1980\*  
*Stewart-Warner Refrigerator Service Manual* Stewart-Warner Corporation,1937      **Stewart-Warner Refrigerator**  
**Service Manual** Stewart-Warner Corporation,1938      **Coldspot and Kenmore Refrigerators Service Manual** Sears,  
Roebuck and Company,1981      **Step-by-step Refrigerator & Freezer Repair Manual** General Electric Company,1990  
Gibson Electric Refrigerator Gibson Electric Refrigerator Corporation,1938\*      **Sears Coldspot and Kenmore**  
**Refrigerators Service Manual** Sears, Roebuck and Company,1981      Service Manual and Parts Catalog, Conventional  
Refrigerators Magic Chef, inc,1979      *Sears Coldspot and Kenmore Refrigerators* Sears, Roebuck and Company,1977  
*Electrolux Service Manual* Electrolux Home Products North America (Firm),2001      *Service Manual, BH Model*  
*Refrigerators* General Electric Company,1959\*      **Norge Master Domestic Refrigerator Service Manual** Norge  
Appliance Service Dept,1952\*

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web lewis structures represent the valence electrons present in the molecule as lone pairs unshared electrons drawn as dots or bonds shared electrons drawn as lines procedure for drawing lewis structures add up the total number of valence electrons that each atom contributes to the molecule polyatomic ion

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web the pcl 3 lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note

the lone pair dots without bonds on top of p just like for n in the previous example for nh 3 chlorine cl

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web step 1 figure out how many electrons the molecule must have based on the number of valence electrons in each atom when drawing the structure of an ion be sure to add subtract electrons to account for the charge step 2 connect the atoms to each other with single bonds to form a skeleton structure

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web this exercise required us to take the formulas in the data table and calculate the number of valence electrons using the periodic table draw its lewis structure on paper build the molecular model and record a photo list the number of bonds and lone pairs surrounding the central atom and lastly identify the structure geometry

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