

Periodic Trends – Atomic Mass

- The **atomic mass** describes the mass of each individual element.
- **Atomic mass** increases moving **down** each group and **right** across a period.

Instructions: In the periodic table below, draw two arrows representing the direction that atomic mass **increases** across groups and periods.

	1																	18
1		2											13	14	15	16	17	
2																		
3			3	4	5	6	7	8	9	10	11	12						
4																		
5																		
6																		
7																		

Instructions: In each question, identify the element with the **greatest** atomic mass:

- 1) Chlorine (Cl), Iodine (I), Bromine (Br), Fluorine (F) Iodine
- 2) Oxygen (O), Carbon (C), Nitrogen (N), Boron (B) Oxygen
- 3) Lithium (Li), Silicon (Si), Sulfur (Br), Xenon (Xe) Xenon

Instructions: In each question, list the elements from **least** to **greatest** atomic mass:

- 4) Magnesium (Mg), Beryllium (Be), Barium (Ba) Be, Mg, Ba
- 5) Gallium (Ga), Selenium (Se), Potassium (K) K, Ga, Se

Periodic Table Trends Answer

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A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Periodic Table Trends Answer

1. Understanding the eBook Periodic Table Trends Answer
 - The Rise of Digital Reading Periodic Table Trends Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Trends Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Trends Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Trends Answer
 - Personalized Recommendations
 - Periodic Table Trends Answer User Reviews and Ratings
 - Periodic Table Trends Answer and Bestseller Lists

5. Accessing Periodic Table Trends Answer Free and Paid eBooks
 - Periodic Table Trends Answer Public Domain eBooks
 - Periodic Table Trends Answer eBook Subscription Services
 - Periodic Table Trends Answer Budget-Friendly Options
6. Navigating Periodic Table Trends Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Trends Answer Compatibility with Devices
 - Periodic Table Trends Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Trends Answer
 - Highlighting and Note-Taking Periodic Table Trends Answer
 - Interactive Elements Periodic Table Trends Answer
8. Staying Engaged with Periodic Table Trends Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Trends Answer
9. Balancing eBooks and Physical Books Periodic Table Trends Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Trends Answer
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Trends Answer
 - Setting Reading Goals Periodic Table Trends Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Trends Answer
 - Fact-Checking eBook Content of Periodic Table Trends Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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