

Stoichiometry Worksheet 2: Percent Yield

For each of the problems below:

- Write the balanced chemical equation
- Identify the given (with units) and what you want to find (with units)
- Show set up with units. Check sig figs, give final answer with units and label.

- Using the Hoffman apparatus for electrolysis, a chemist decomposes 36 g of water into its gaseous elements. How many grams of hydrogen gas should she get (theoretical yield)?



Before **2 mol** **0 mol** **0 mol**

Change **-2 mol** **+2 mol**

After **0 mol** **2 mol**

$$36 \text{ g H}_2\text{O} \times \frac{1 \text{ mol H}_2\text{O}}{18.0 \text{ g H}_2\text{O}} = 2.0 \text{ mol H}_2\text{O} \times \frac{2 \text{ mol H}_2}{2 \text{ mol H}_2\text{O}} = 2 \text{ mol H}_2 \times \frac{2.0 \text{ g H}_2}{1 \text{ mol H}_2} = 4.0 \text{ g H}_2$$

- Recall that liquid sodium reacts with chlorine gas to produce sodium chloride. You want to produce 581 g of sodium chloride. How many grams of sodium are needed?



Before **9.94 mol** **xs mol** **0 mol**

Change **-9.94 mol** **+9.94 mol**

After **0 mol** **9.94 mol**

$$581 \text{ g NaCl} \times \frac{1 \text{ mol NaCl}}{58.5 \text{ g NaCl}} = 9.93 \text{ mol NaCl} \times \frac{2 \text{ mol Na}}{2 \text{ mol NaCl}} = 9.93 \text{ mol Na} \times \frac{23.0 \text{ g Na}}{1 \text{ mol Na}} = 228 \text{ g Na}$$

- You eat 180.0 g of glucose (90 M&Ms). If glucose, $\text{C}_6\text{H}_{12}\text{O}_6$, reacts with oxygen gas to produce carbon dioxide and water, how many grams of oxygen will you have to

Percent Yield Stoichiometry Answers With Work

RS Peters



Percent Yield Stoichiometry Answers With Work:

Embracing the Song of Phrase: An Mental Symphony within **Percent Yield Stoichiometry Answers With Work**

In a global eaten by displays and the ceaseless chatter of quick conversation, the melodic elegance and mental symphony produced by the prepared term often diminish in to the back ground, eclipsed by the relentless sound and disturbances that permeate our lives. However, situated within the pages of **Percent Yield Stoichiometry Answers With Work** an enchanting literary prize full of natural thoughts, lies an immersive symphony waiting to be embraced. Crafted by a wonderful composer of language, that interesting masterpiece conducts viewers on a psychological journey, well unraveling the concealed songs and profound impact resonating within each carefully constructed phrase. Within the depths of this poignant evaluation, we will explore the book is key harmonies, analyze its enthralling publishing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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