

Molecular Geometry Chart

# of Electron Groups	Number of Lone Pairs	Electron Pair Arrangement	Molecular Geometry	Approximate Bond Angles
2	0	linear		180°
3	0	trigonal planar		120°
	1	bent		<120°
4	0	tetrahedral		109.5°
	1	trigonal pyramid		<109.5° (~107°)
	2	bent		<109.5° (~105°)
5	0	trigonal bipyramidal		90°, 120°
	1	see-saw		<90°, <120°
	2	T-structure		<90°
	3	linear		180°
6	0	octahedral		90°, 90°
	1	square pyramidal		90°, <90°
	2	square planar		90°

Shapes Of Molecules Worksheet If8766

Dale Atkins

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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The Shape and Structure of Molecules Charles Alfred Coulson, 1982 *Molecular Geometry* Alison Rodger, Mark Rodger, 1995 Molecular Geometry discusses topics relevant to the arrangement of atoms The book is comprised of seven chapters that tackle several areas of molecular geometry **Molecules and Models** Arne Haaland, 2008-03-06 This book provides a systematic description of the molecular structures and bonding in simple compounds of the main group elements with particular emphasis on bond distances bond energies and coordination geometries The description includes the structures of hydrogen halogen and methyl derivatives of the elements in each group some of these molecules are ionic some polar covalent The survey of molecules whose structures conform to well established trends is followed by representative examples of molecules that do not conform We also describe electron donor acceptor and hydrogen bonded complexes Chemists use models to systematize our knowledge to memorize information and to predict the structures of compounds that have not yet been studied The book provides a lucid discussion of a number of models such as the Lewis electron pair bond and the VSEPR models the spherical and polarizable ion models and molecular orbital calculations and it outlines the successes and failures of each Shape in Chemistry Paul G. Mezey, 1993 Molecules Dale Atkins, 1987-08-15 Explains how diagrams are used to represent chemical bonds and describes the structure and characteristics of molecules encountered in everyday life

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