

Chemical Periodicity

Name Mitchell Wade Section Number 015

OBJECTIVE: To demonstrate the periodic law.

CONCEPTS: Periodic law, Periodic trends

INTRODUCTION: In the 1860's about 60 elements were known. From their physical properties, the elements were classified as metals, nonmetals, or metalloids. Metals have low electrical resistance (high conductivity), have a metallic luster, and are malleable. Non-metals have high resistance, low luster, and are brittle. Metalloids have properties between those of non-metals and metals. Chemists of the time further classified the elements into groups. The elements in a particular group have similar chemical and/or physical properties. This classification led to the discovery of the periodic table and the periodic law. In the periodic table, elements are assigned to a period and a group—groups are the columns and the periods are the rows of the table. The periodic law states that the properties of the chemical elements vary with the atomic number in a systematic way. We now understand that the properties depend on the electronic structures of the atoms.

PROCEDURE:

1. Find the station labeled "Halogens: states of matter." Observe the states of matter of the three samples. How does the state of matter change as you go down the column of the periodic table?

Starts as gas and goes to solid

The melting points of the halogen elements that you observed are listed below. Based on the trend that you see, estimate the melting point of Fluorine.

Element	Melting Point ($^{\circ}\text{C}$)
F	-200
Cl	-101.5
Br	-7.35
I	113.75

2. Find the station labeled "Metallic Solutions." Fill out the chart below. For the unlabeled mystery elements, you will have to guess which type of metal they are (transition, alkali, or alkaline earth) based on the trends you see.

Solution Number	Element	Color of Water Solution	Type (transition metal, alkali metal, alkaline earth metal)
1	Co	Red	Transition metal
2	Ba	Clear	Alkaline earth metal
3	K	clear	Alkali metal
4	Ni	green	Transition metal
5	Na	clear	Alkali metal
6	Cu	blue	Transition metal
7	Mg	clear	Alkaline earth metal

Periodicity Lab Report

Michael Seilmaier



Periodicity Lab Report:

Scientific and Technical Aerospace Reports ,1994 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database **U.S. Government Research Reports** ,1964 Report of the Annual Meeting British Association for the Advancement of Science. Meeting,1920 *Sustainable Green Chemistry* Mark Anthony Benvenuto,2017-04-10

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web sep 8 2023 organelles are small structures within the cytoplasm that carry out functions necessary to maintain homeostasis in the cell they are involved in many processes for example energy production building proteins and secretions destroying toxins and responding to external signals

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web 4 transports materials within the cell vesicles 5 the region inside the cell except for the nucleus cytoplasm 6 organelle that manages or controls all the cell functions in a eukaryotic cell nucleus 7 contains chlorophyll a green pigment that traps energy from sunlight and gives plants their green color chloroplast 8 digests excess or

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