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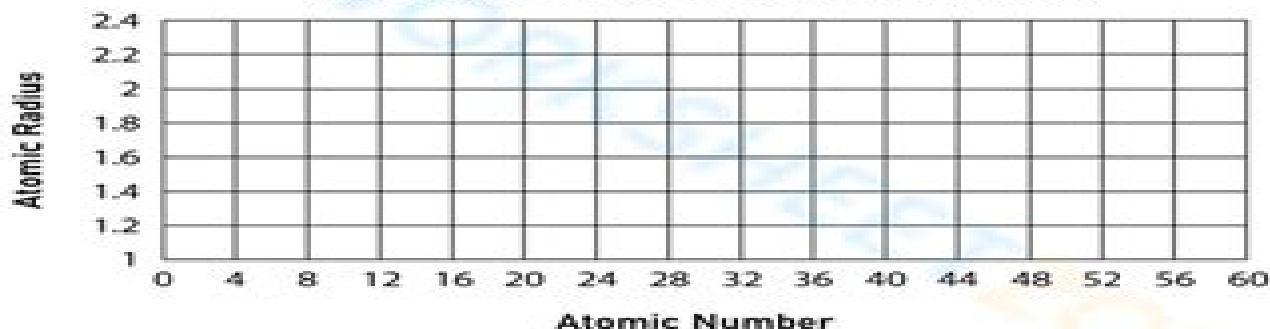
## Graphing Periodic Trends Worksheet

Using the data below, make line graphs of atomic radius vs. atomic number for Group 2A and Period 3 of the periodic table.

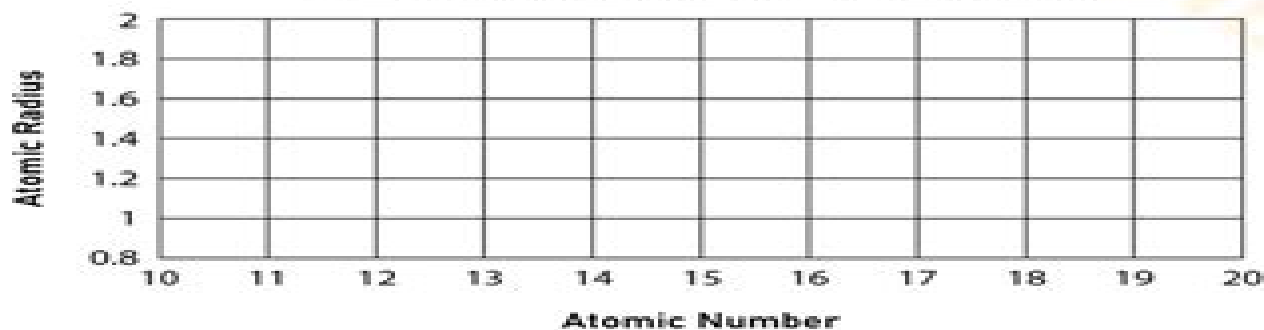
| Group 2A |               |               |
|----------|---------------|---------------|
| Element  | Atomic Number | Atomic Radius |
| Be       | 4             | 1.11          |
| Mg       | 12            | 1.60          |
| Ca       | 20            | 1.97          |
| Sr       | 38            | 2.15          |
| Ba       | 56            | 2.17          |

| Period 3 |               |               |
|----------|---------------|---------------|
| Element  | Atomic Number | Atomic Radius |
| Na       | 11            | 1.86          |
| Mg       | 12            | 1.60          |
| Al       | 13            | 1.43          |
| Si       | 14            | 1.17          |
| P        | 15            | 1.10          |
| S        | 16            | 1.04          |
| Cl       | 17            | 0.99          |
| Ar       | 18            | 0.94          |

Group 2A: Atomic Radius vs. Atomic Number



Period 3: Atomic Radius vs. Atomic Number



# Periodic Trends Answer

**M Carnoy**



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