

# Periodic Trends – Atomic Mass

- The **atomic mass** describes the mass of each individual element.
- **Atomic mass** increases moving **down** each group and **right** across a period.

**Instructions:** In the periodic table below, draw two arrows representing the direction that atomic mass **increases** across groups and periods.

|   |   |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    |    |  |
|---|---|---|---|---|---|---|---|---|---|----|----|----|--|----|----|----|----|----|--|
|   | 1 |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    | 18 |  |
| 1 |   | 2 |   |   |   |   |   |   |   |    |    |    |  | 13 | 14 | 15 | 16 | 17 |  |
| 2 |   |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    |    |  |
| 3 |   |   | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 |  |    |    |    |    |    |  |
| 4 |   |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    |    |  |
| 5 |   |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    |    |  |
| 6 |   |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    |    |  |
| 7 |   |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    |    |  |

**Instructions:** In each question, identify the element with the **greatest** atomic mass:

- 1) Chlorine (Cl), Iodine (I), Bromine (Br), Fluorine (F) Iodine
- 2) Oxygen (O), Carbon (C), Nitrogen (N), Boron (B) Oxygen
- 3) Lithium (Li), Silicon (Si), Sulfur (Br), Xenon (Xe) Xenon

**Instructions:** In each question, list the elements from **least** to **greatest** atomic mass:

- 4) Magnesium (Mg), Beryllium (Be), Barium (Ba) Be, Mg, Ba
- 5) Gallium (Ga), Selenium (Se), Potassium (K) K, Ga, Se

# Periodic Trends Worksheet Extension Questions

**Joann H. Lee**



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## **Table of Contents Periodic Trends Worksheet Extension Questions**

1. Understanding the eBook Periodic Trends Worksheet Extension Questions
  - The Rise of Digital Reading Periodic Trends Worksheet Extension Questions
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Worksheet Extension Questions
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Trends Worksheet Extension Questions
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Worksheet Extension Questions
  - Personalized Recommendations
  - Periodic Trends Worksheet Extension Questions User Reviews and Ratings
  - Periodic Trends Worksheet Extension Questions and Bestseller Lists
5. Accessing Periodic Trends Worksheet Extension Questions Free and Paid eBooks
  - Periodic Trends Worksheet Extension Questions Public Domain eBooks
  - Periodic Trends Worksheet Extension Questions eBook Subscription Services
  - Periodic Trends Worksheet Extension Questions Budget-Friendly Options
6. Navigating Periodic Trends Worksheet Extension Questions eBook Formats

- ePub, PDF, MOBI, and More
  - Periodic Trends Worksheet Extension Questions Compatibility with Devices
  - Periodic Trends Worksheet Extension Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Periodic Trends Worksheet Extension Questions
    - Highlighting and Note-Taking Periodic Trends Worksheet Extension Questions
    - Interactive Elements Periodic Trends Worksheet Extension Questions
  8. Staying Engaged with Periodic Trends Worksheet Extension Questions
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Periodic Trends Worksheet Extension Questions
  9. Balancing eBooks and Physical Books Periodic Trends Worksheet Extension Questions
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Periodic Trends Worksheet Extension Questions
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Periodic Trends Worksheet Extension Questions
    - Setting Reading Goals Periodic Trends Worksheet Extension Questions
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Periodic Trends Worksheet Extension Questions
    - Fact-Checking eBook Content of Periodic Trends Worksheet Extension Questions
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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